## **CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**International General Certificate of Secondary Education** 

## MARK SCHEME for the May/June 2014 series

## 0654 CO-ORDINATED SCIENCES

**0654/63** Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2014 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.



	Page 2			Mark Scheme	Syllabus	Paper	
				IGCSE – May/June 2014	0654	63	
1	(a)	(i)		d quality drawing ; el, stamen and at least two petals drawn ;		[2]	
		(ii)		nen correctly labelled ; el correctly labelled ;		[2]	
	(b)	(i)	corre	ect measurement from photograph – 68 (mm) ;		[1]	
		(ii)	corre	ect measurement of drawing given ;		[1]	
	(c)	magnification calculated by dividing the length of drawing by the length of the petal in photo (ensure both in the same units);					
	(d)	stigma labelled <b>Z</b> ;					
	(e)	select <u>anther</u> (allow top of stamen); squash/cut to open anther;					
		use	a mi	croscope to observe;		[max 2]	
						[Total: 10]	
2	(a)	(i)	<b>A</b> an	nd <b>F</b> (both required, either order);		[1]	
		(ii)		oles with sodium carbonate ; eaction with hydrochloric acid ;		[2]	
	(b)	copper(II) chloride: blue ppt;					
	( - )	becomes (dark) blue solution ;		ro.			
		aqueous ammonia: no change / no reaction ;			[3]		
	(c)	(i)	no o	bservable change/no ppt ;		[1]	
		(ii)	sulfu				
				sulfate: no change/no ppt; um chloride: white ppt;		[3]	
				o marks is hydrochloric acid is used)		[~]	
						[Total: 10]	
3	(a)	73.5 71.0				[2]	
		71. <u>0</u> ;					
	(b)	axes correct and labelled and use of grid;					
		points (allow 1 error); smooth curve;				[3]	
		51110		,		[0]	

Page 3		Mark S		Syllabus	Paper
•	IGCSE – May/June 2014			0654	63
(c) (i)	two figures fro	om graph/90 ; ng ;			[2]
(ii)	value less tha	n <b>(i)</b> ;			[1]
` '	rnal temperati		ter/volume of water ;		
	rial of beaker	•			[max 2]
					[Total:10]
(a) incr	eases ;				[1]
(b) (i)					
	pulse rate/be	ats per min			
	104				
	80				
	72				
	(3 correct = 2	marks, 2 correc	t = 1 mark)		[max 2]
(ii)	<i>beats</i> = 256 ;				[1]
	93.75/94/93. ss rating: exce				[2]
(d) (i)	twin <b>A</b> : 400 <b>A</b>	<b>ND</b> <i>twin</i> <b>B</b> : 393 ;	i		
(ii)	twin <b>A</b> : poor <b>A</b>	AND <i>twin</i> B: aver	rage ;		[1]
(iii)	experimental arbitrary cut o	ff;	wtte ; ute in heart rate ;		
	AVP ;	n minute to milit	ac iii iicait iate ,		[max 2]
					[Total: 10]

		IGCSE – May/June 2014	0654	63
5	(a) (i)	axes correct and labelled; use of grid; points (allow 1 error); curve;		[4]
	(ii)	from candidate's graph (about 15); accuracy/extrapolation;		[2]
	(iii)	lowers it;		[1]
	(iv)	from graph 132 – 42 (marking on candidate's graph) = 90 ;		[2]
	(b)	slower process/heating at one position;		[1]
				[Total: 10]
6	(a) (i)	ammeter in series ; voltmeter in parallel ; correct cell ;		[3]
	(ii)	A = 0.35; V = 1.55;		[2]
	(iii)	resistance = 4.43 ; (ecf) unit = $\Omega$ (allow ohm) ;		[2]

(b) (i) (ammeter reading) decreases AND (brightness) not as bright/dimmer

(both required);

(ii) brighter as more current flows;

then 'blows' as filament melts;

**Mark Scheme** 

**Syllabus** 

**Paper** 

[Total: 10]

[1]

[2]

Page 4